



Experiment Report Form



Experiment title: XAFS study of nanoparticle catalysts in H-sorption kinetics of Mg based nanocomposites	Experiment number: ME-1291	
Beamline:	Date of experiment: from: 6 th April 2006 to: 11 th April 2006	Date of report: 11-08-2007
Shifts:	Local contact(s): Dr. Gemma Guilera	<i>Received at ESRF:</i>
Names and affiliations of applicants (* indicates experimentalists): Prof. Asunción Fernández*, Instituto de Ciencia de Materiales de Sevilla Dr. Oliver Friedrichs*, Instituto de Ciencia de Materiales de Sevilla Dr. Diego Martínez*, Instituto de Ciencia de Materiales de Sevilla		

Report:

10700

J. Phys. Chem. C 2007, 111, 10700–10706

In Situ Energy-Dispersive XAS and XRD Study of the Superior Hydrogen Storage System MgH₂/Nb₂O₅

Oliver Friedrichs,^{*,†,§} Diego Martínez-Martínez,[†] Gemma Guilera,[‡]
 Juan Carlos Sánchez López,[†] and Asunción Fernández[†]

Instituto de Ciencia de Materiales de Sevilla (CSIC–Universidad Sevilla), Avenida Américo Vespucio 49, 41092 Sevilla, Spain, and European Synchrotron Radiation Facility, F-38043 Grenoble 09, France

Received: November 15, 2006; In Final Form: May 3, 2007

By in situ energy-dispersive X-ray absorption spectroscopy (EDXAS) and X-ray diffraction (XRD), we analyzed the evolution of niobium in a MgH₂/Nb₂O₅ system based on high-energy ball milling during hydrogen cycling. The high time resolution of the EDXAS method allowed us to monitor fast sample changes during this process. Thereby, we demonstrated that the Nb₂O₅ is already partially reduced during the milling process with the MgH₂. Further reduction occurs during the heating and cycling processes, in which a lower limit of oxidation state is reached. Hereby, a reaction between the niobium oxide and the Mg/MgH₂ leads to a decrease of crystalline Nb₂O₅ and the formation of a ternary oxide phase Mg_xNb_yO. During the cycling processes a repetitive Nb oxidation–reduction process was observed, which may indicate hydrogen diffusion along the ternary oxide by the formation of metastable niobium hydrides. This points to a mechanism of kinetic sorption improvement by diffusion of hydrogen through pathways of ternary Mg–Nb oxides, which may also reduce the activation energy of the Mg–MgH₂ transition.

