 ROBL-CRG	Experiment title: In-situ XRD/XRR of thin film catalysis during carbon nanotube growth	Experiment number: 20_02_671
Beamline: BM 20	Date of experiment: from: 21.10.2008-28.10.2008	Date of report: 6.5.2009
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Names and affiliations of applicants (* indicates experimentalists): C. Baehtz*, T. Wirth ^{1)*} , C.S. Esconjauregui ^{1)*} Forschungszentrum Rossendorf, Institute of Ion Beam Physics and Materials Research, P.O.B. 510119, 01314 Dresden, Germany 1) Centre for Advanced Photonics and Electronics, University of Cambridge, 9 JJ Thompson Avenue, Cambridge CB3 0FA, UK		

Results

Metal nano particles like Fe or Ni can act as catalyst in the carbon nano tube (CNT) growth process [1,2]. These metals were first deposited as thin film onto Si-wafer with SiO₂ or Al₂O₃ buffer layers and then heated up under inert or reducing atmosphere. Hereby the film dewets and the nano particles are formed. The crystal sizes of the

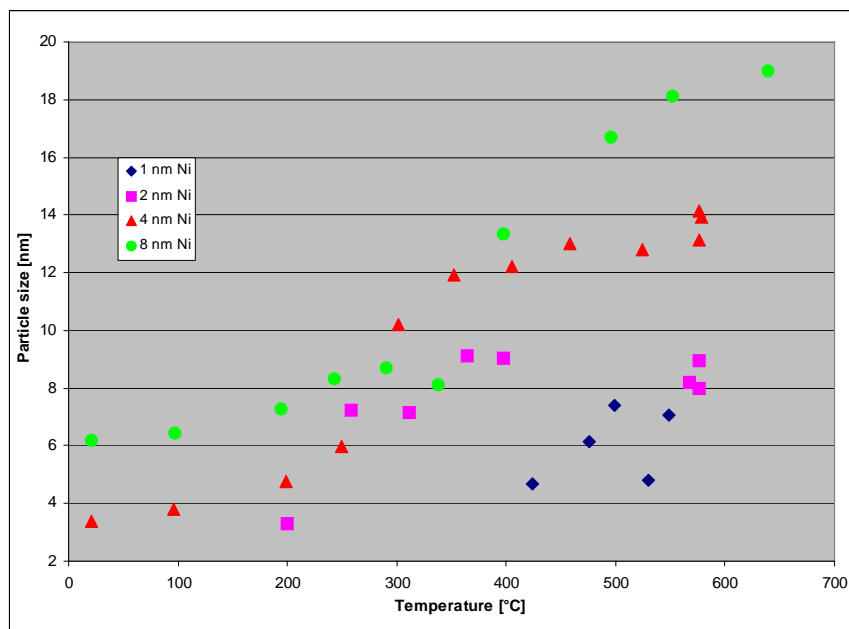


Fig.1: Temperature dependance of the Ni nano-crystallite sizes during heating and the subsequent dewetting.

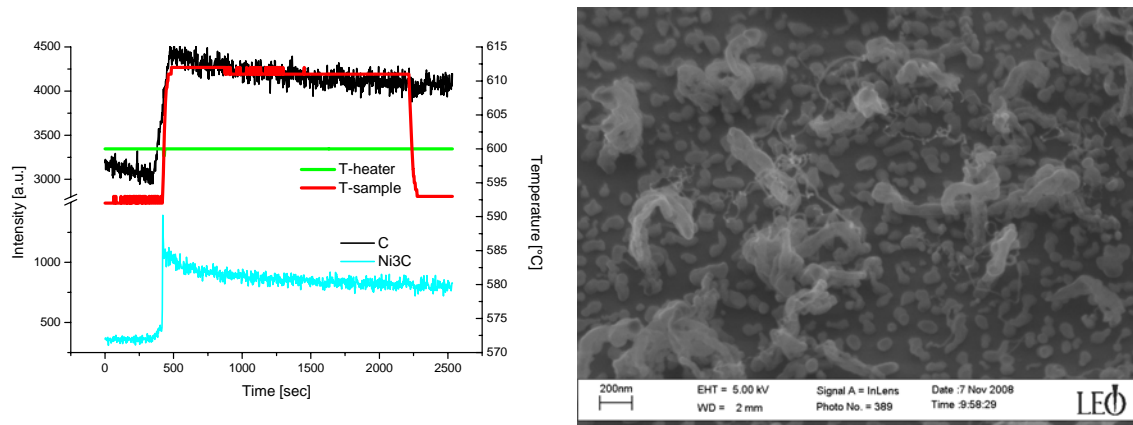


Fig. 2: Timereolved measurement of the Ni_3C and C signal and the corresponding SEM picture (right).

resulting nano-particle depend on the thickness of the initial film. A crude approximation shows that the particles became three times bigger than in the film.

During the growth experiment using 2 or 4 nm thick Ni film system a second phase Ni_3C is observed. This raises the question if this could be the catalytic active material. Time resolved XRD in Figure 2 monitoring the Ni_3C and graphite signal shows the Ni_3C formation directly after the H_2C_2 injection, which act as carbon source. With a slight delay the C signal also increases. Both signal are stable after approx. 10 min. The increased sample temperature during the H_2C_2 feed indicates an exothermic reaction. Based on the corresponding SEM picture the C-signal can be attributed to the low yield CNT but also on surface coking carbon. Further experiments to elucidate this aspect were planed in future.

References

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- [2] In-situ observations of catalyst dynamics during surface bond carbon nanotube nucleation, Hofmann S., Sharma R., Ducati C., Du G., Mattevi C., Cepek C., Cantoro M., Pisana S., Parvez A., Ferrari A. C., Dunin-Borkowski R., Lizzit S., Petaccia L., Goldoni A., Robertson J., Nano Lett. 7, 602 (2007).