




Experiment Report Form

	Experiment title: Stability of chlorinated RuO ₂ (110) model catalyst in the Sumitomo Process	Experiment number: SI-1944
	Beamline: ID03	Date of experiment: from: 23.9.2009 to: 29.9.2009
Shifts: 18	Local contact(s): Dr. Olivier Balmes	<i>Received at ESRF:</i>
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Report: submitted to Journal of Catalysis: Paper has been accepted on 16.2.2010.

In-situ Studies of the Oxidation of HCl over RuO₂ Model Catalysts:

Stability and Reactivity

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Abstract

Structure-activity experiments were performed for the HCl oxidation reaction (Deacon-like process) over RuO₂ model catalysts – chlorinated RuO₂(110) and RuO₂(100) - applying in-situ surfaces x-ray diffraction (SXRD) combined with on-line mass spectrometry. The studied model catalysts turned out to be long-term

stable under reaction conditions with gas feed ratios $p(\text{HCl}):p(\text{O}_2)$ ranging from 1:4 to 4:1 in the mbar pressure regime and temperatures as high as 685 K. Even pure HCl exposure in the mbar regime was not able to reduce RuO_2 below 600 K; above 650 K chemical reduction of the oxide sets in. Under strongly oxidizing reaction conditions the (surface) oxides grow slowly in thickness. On-line reactivity experiments of both types of model catalysts in a batch reactor yield a mean turn-over frequency (TOF) of 0.6 Cl_2 molecules per second and active site for the HCl oxidation at 650 K and initial partial pressures of $p(\text{HCl}) = 2$ mbar and $p(\text{O}_2) = 0.5$ mbar. The HCl-oxidation over RuO_2 is therefore considered to be structure insensitive.

Graphical Abstract:

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In-situ surface x-ray diffraction reveals that RuO_2 -model catalysts are long-term stable for the HCl oxidation reaction by oxygen with a mean TOF of 0.6 Cl_2/s using a batch reactor.

