



	Experiment title: Metal borohydrides with perovskite structure type: ionic conduction and hydrogen storage.	Experiment number: 01-02-1037
Beamline: BM01-A	Date of experiment: from: January 29 to: February 1, 2014	Date of report: September 2014
Shifts: 9	Local contact(s): Dr. V. Diadkin	<i>Received at ESRF:</i>
Names and affiliations of applicants (* indicates experimentalists): *R. Cerny, Laboratoire de Cristallographie, Geneve, Switzerland P. Schouwink, Laboratoire de Cristallographie, Geneve, Switzerland *Y. Sadikin, Laboratoire de Cristallographie, Geneve, Switzerland *Ya. Filinchuk, Université de Louvain, Belgium *Iu. Dovgaliuk, Université de Louvain, Belgium *K. Robeyns, Université de Louvain, Belgium *G. Springuel, Université de Louvain, Belgium		

Many different systems containing perovskite borohydrides $AB(\text{BH}_4)_3$ with A = alkali metal, B = rare-earth were studied. Two typical examples are given below:

LiBH₄ + KBH₄ + YbCl₃

The perovskite phase $\text{KYb}(\text{BH}_4)_3$ is formed in ball milled mixtures 12:4:3. At RT the perovskite is simple cubic, and at 95°C it transforms to the orthorhombic phase with $8 \times$ superstructure $\text{SQRT}(2) / 2 / 2\text{SQRT}(2)$ which was not reported among oxide perovskites [1]:

Compound	SG	$V(\text{Å}^3)$	$V/Z(\text{Å}^3)$	$a(\text{Å})$	$b(\text{Å})$	$c(\text{Å})$	T/K
LT-KYb(BH ₄) ₃	$P-43m$	176.3	176.3	5.607			298
HT-KYb(BH ₄) ₃	$Pm2_1b$	1423.8	177.9	7.832	11.267	16.135	403

Evolution of unit cell volume and lattice parameters with temperature is shown on Figure 1. The decrease of the cell volume above 180°C is due to dissolution of chloride from LiCl in the perovskite.

Lanthanides form the basis of RGB (red-green-blue) colours in LED-based solid-state lighting with many important applications. Room temperature emission and excitation

spectra of $\text{KYb}(\text{BH}_4)_3$ were measured in our laboratory down to 10 K. The Yb^{2+} emission is centred at 520 nm, when excited in the UV at 355 nm [1].

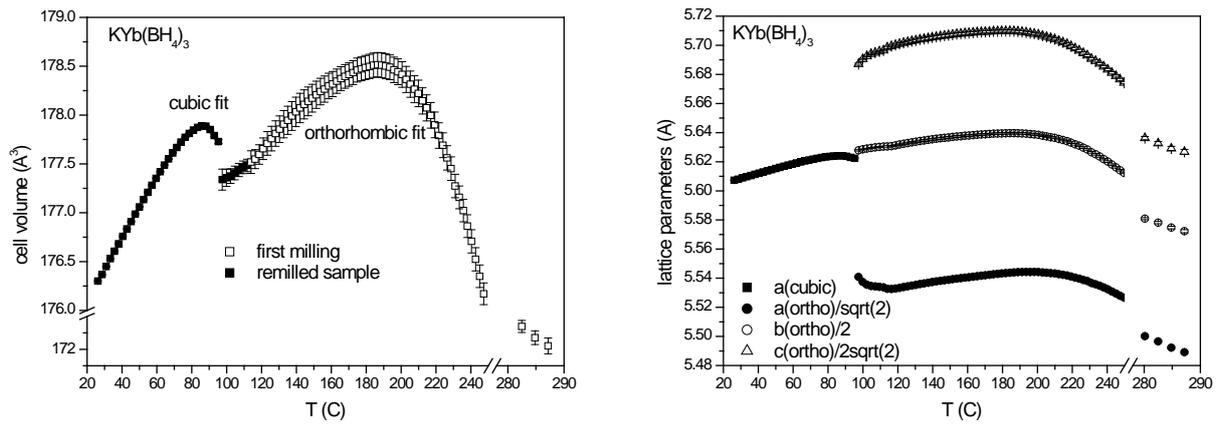


Figure 1: Evolution of unit cell volume and lattice parameters with temperature for cubic and orthorhombic perovskite $\text{KYb}(\text{BH}_4)_3$.

$\text{LiBH}_4 + \text{KBH}_4 + \text{Gd}(\text{BH}_4)_3$

Gadolinium crystals are of interest as host materials for trivalent lanthanides based phosphors. Gadolinium forms double-perovskite phases $\text{Cs}_3\text{Gd}(\text{BH}_4)_6$ and $\text{Cs}_2\text{LiGd}(\text{BH}_4)_6$ with bigger alkali metal [1]. A different phase is formed with smaller alkali metal K in the mixtures 4:3:3, but also without lithium in mixtures 0:3:2 as shown on the T-ramp of 4:3:3 mixture in Figure 2.

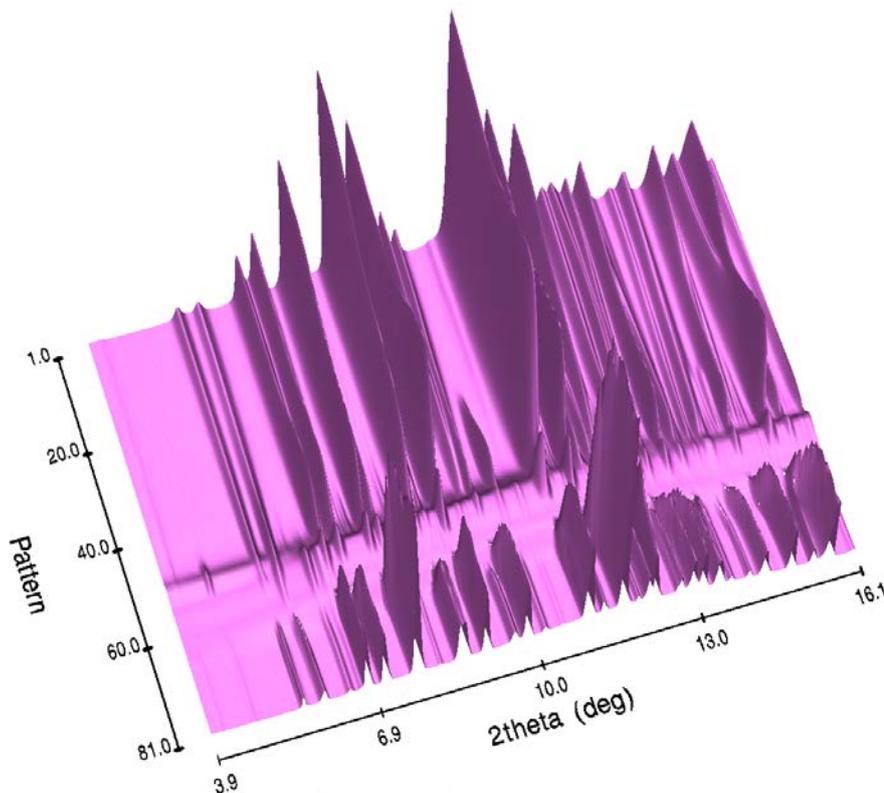


Figure 2: T-ramp of the ball-milled mixture $4 \text{LiBH}_4 + 3 \text{KBH}_4 + 3 \text{Gd}(\text{BH}_4)_3$ showing a new phase formed by ball milling at RT, and disappearing in the pattern no. 53. at $T \sim 150^\circ\text{C}$.

Despite a great effort invested in its crystal structure solution, no satisfactory model has been obtained yet.

[1] Schouwink et al., *Nature Comm.*, submitted.