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|                           | <b>Experiment title:</b><br>Monitoring bio-inspired cuboidal $\text{Co}_4\text{O}_4$ water oxidation catalysts at work with XAS techniques | <b>Experiment number:</b><br>01-01-961 |
| <b>Beamline:</b><br>BM01b | <b>Date of experiment:</b><br>from: Oct 29 <sup>th</sup> 2014 to: Nov 3 <sup>rd</sup> 2014   | <b>Date of report:</b><br>25.02.2015   |
| <b>Shifts:</b><br>12      | <b>Local contact(s):</b> Dr. Hermann Emerich   |  |

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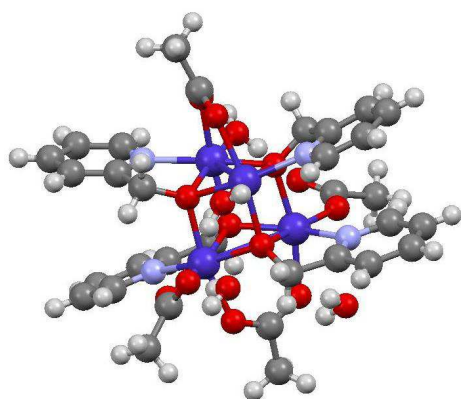
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**Report:**

Photocatalytic watersplitting is an elegant way to feed the growing global energy demand with green and sustainable fuels.<sup>[1]</sup> However, the water oxidation half-reaction remains as a four electron process the bottleneck in the overall catalytic cycle. We investigated biometric

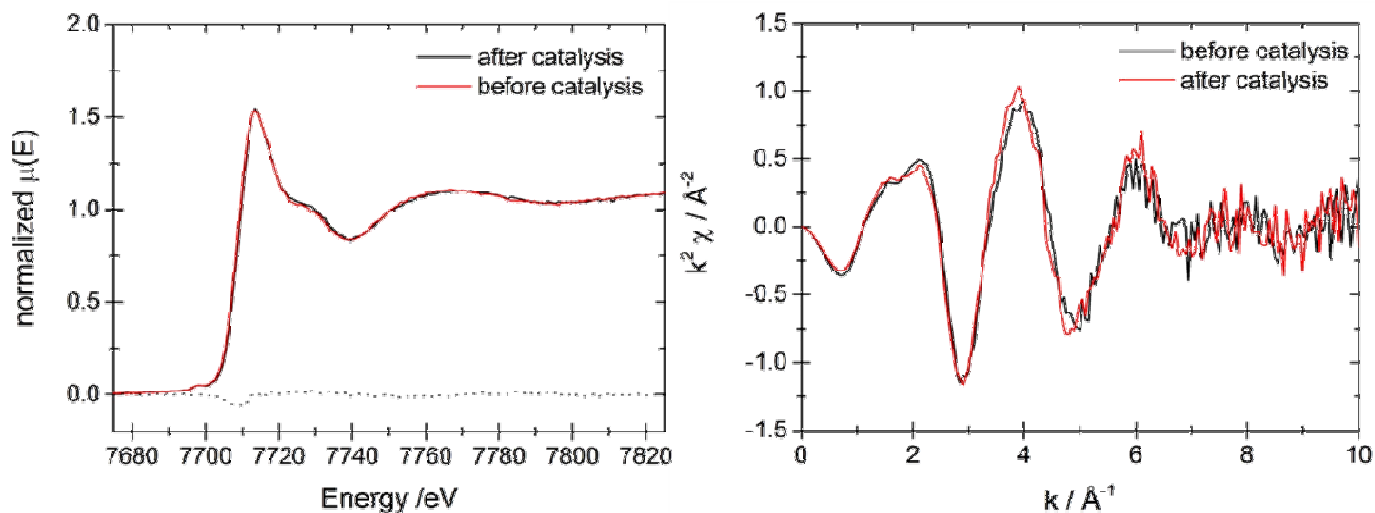


**Fig. 1.** Molecular structure of the water oxidation catalyst  $\text{Co}_4\text{O}_4$ .

cuboidal water oxidation catalyst ( $\text{Co}_4\text{O}_4$ ,  $[\text{Co}_4(\text{hmp})_4(\text{OAc})_4(\text{H}_2\text{O})_2]$ , hmp = 2-Methylhydroxypyridine, OAc = acetyl) (Fig 1.).<sup>[2]</sup>

An average valance state of +2 could be estimated for  $\text{Co}_4\text{O}_4$  in the catalytic solution based on the edge position in respect to known reference standards. The comparison of the XANES region of  $\text{Co}_4\text{O}_4$  in catalytic solution before and after photocatalysis displays no of the shift edge position and no significant change in the difference spectra (Fig. 2 right). Therefore, any kind of irreversible oxidation of  $\text{Co}_4\text{O}_4$  during photocatalyst can be excluded.

In line with the nearly identical EXAFS region of  $\text{Co}_4\text{O}_4$  before and after catalysis (Fig. 2 left), changes in the coordination enviroment of the Co can be neglected.



**Fig. 2.** Left: XANES region and difference spectrum of  $\text{Co}_4\text{O}_4$  before and after catalysis. Right: EXAFS region of 1 before and after catalysis plotted in K-space.

### References:

- [1] H. Lv, Y. Geletii, C. Zhao, J. W. Vickers, G. Zhu, Z. Luo, J. Song, T. Lian, D. G. Musaev, C. L. Hill, *Chem. Soc. Rev.*, 2012, **41**, 7572.
- [2] F. Evangelisti, R. Güttinger, R. Moré, S. Luber, G.R. Patzke, *J Am. Chem. Soc.*, 2013, **135**, 18734.