



	<b>Experiment title:</b> In situ total scattering studies on alkali metal nitrate promoted MgO absorbents for CO <sub>2</sub> capture at moderate temperatures	<b>Experiment number:</b> MA-3415
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<b>Names and affiliations of applicants</b> (* indicates experimentalists): Abdala Paula Macarena* <sup>1</sup> , Dal Pozzo Alessandro* <sup>1</sup> , Rekhtina Margarita* <sup>1</sup> , Jakub Drnec* <sup>2</sup> , Blanco Maria* <sup>2</sup> and Christoph Müller <sup>1</sup> <sup>1</sup> Laboratory Energy Technology Laboratory of Energy Science and Engineering Leonhardstrasse 21, LEE P201 CH - 8092 ZUERICH <sup>2</sup> Laboratory ESRF 71 avenue des Martyrs CS 40220 FR - 38043 GRENOBLE Cedex 9		

#### Report:

The experiments led to the following publication:

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#### **Effect of molten sodium nitrate on the decomposition pathways of hydrated magnesium hydroxycarbonate to magnesium oxide probed by in situ total scattering†**

Margarita Rekhtina<sup>a</sup>, Alessandro Dal Pozzo<sup>ab</sup>, Dragos Stoian<sup>c</sup>, Andac Armutlulu<sup>a</sup>, Felix Donat<sup>a</sup>, Maria V. Blanco<sup>c</sup>, Zhu-Jun Wang<sup>d</sup>, Marc-Georg Willinger<sup>d</sup>, Alexey Fedorov<sup>a</sup>, Paula M. Abdala<sup>\*a</sup> and Christoph R. Müller<sup>\*a</sup>

<sup>a</sup>Laboratory of Energy Science and Engineering, Department of Mechanical and Process Engineering, ETH Zürich, Leonhardstrasse 21, 8092 Zürich, Switzerland. E-mail: [abdalap@ethz.ch](mailto:abdalap@ethz.ch); [muelchri@ethz.ch](mailto:muelchri@ethz.ch)

<sup>b</sup>Laboratory of Industrial Safety and Environmental Sustainability, Department of Civil, Chemical,

Environmental and Materials Engineering, Alma Mater Studiorum–Università di Bologna, Via Terracini 28,  
40131 Bologna, Italy

<sup>c</sup>European Synchrotron Radiation Facility, 71 Avenue des Martyrs, Grenoble, France

<sup>d</sup>Scientific Center for Optical and Electron Microscopy, ETH Zürich, Auguste-Piccard-Hof 1, 8093 Zurich,  
Switzerland

## **Abstract**

The effect of  $\text{NaNO}_3$  and its physical state on the thermal decomposition pathways of hydrated magnesium hydroxycarbonate (hydromagnesite, HM) towards  $\text{MgO}$  was examined by in situ total scattering. Pair distribution function (PDF) analysis of these data allowed us to probe the structural evolution of pristine and  $\text{NaNO}_3$ -promoted HM. A multivariate curve resolution alternating least squares (MCR-ALS) analysis identified the intermediate phases and their evolution upon the decomposition of both precursors to  $\text{MgO}$ . The total scattering results are discussed in relation with thermogravimetric measurements coupled with off-gas analysis.  $\text{MgO}$  is obtained from pristine HM ( $\text{N}_2$ ,  $10\text{ }^\circ\text{C min}^{-1}$ ) through an amorphous magnesium carbonate intermediate (AMC), formed after the partial removal of water of crystallization from HM. The decomposition continues via a gradual release of water (due to dehydration and dehydroxylation) and, in the last step, via decarbonation, leading to crystalline  $\text{MgO}$ . The presence of molten  $\text{NaNO}_3$  alters the decomposition pathways of HM, proceeding now through AMC and crystalline  $\text{MgCO}_3$ . These results demonstrate that molten  $\text{NaNO}_3$  facilitates the release of water (from both water of crystallization and through dehydroxylation) and decarbonation, and promotes the crystallization of  $\text{MgCO}_3$  and  $\text{MgO}$  in comparison to pristine HM.  $\text{MgO}$  formed from the pristine HM precursor shows a smaller average crystallite size than  $\text{NaNO}_3$ -promoted HM and preserves the initial nano-plate-like morphology of HM.  $\text{NaNO}_3$ -promoted HM was decomposed to  $\text{MgO}$  that is characterized by a larger average crystallite size and irregular morphology. Additionally, in situ SEM allowed visualization of the morphological evolution of HM promoted with  $\text{NaNO}_3$  at a micrometre scale.

A second manuscript is currently in progress.