# Preliminary report on experiment CH6424

#### Summary

Four samples were measured, labelled as **tetrazol**, **MePy2**, **dabco** and **quinone**. Diffraction data for **MePy2** and **dabco** were measured up to 0.45 Å and were of high quality, so the charge densities were successfully refined (see below, Figs. 2-10). Crystals of **tetrazol** were small and twinned, which allowed only spheric refinement. Nevertheless, the atomic-resolution crystal structure is very accurate and publicable (see below, Table 1 and Fig. 1). Six high-resolution diffraction data of **quinone** were collected 0.45 Å, and spheric refinement was very good, yielding publicable crystal structures. However, multipolar refinement was difficult and residual density remains unacceptably high. We are still working on multipolar refinement; in the worst case, atomic-resolution crystal structure will be publicable.

#### **Experimental**

Single-crystal X-ray diffraction data were collected at the Rossendorf Beamline (ESRF / Grenoble, France) (Scheinost *et al.*, 2020) equipped with a Si(111) monochromator and two Pd coated mirrors. The single-crystal data were recorded with a Pilatus3 X 2M detector (Dectris) with an excitation energy of 20000 eV / 0.6200926 Å. The monochromator energy was calibrated against the first inflection of the *K*-absorption edge of a Mo metal foil point, tabulated as 20000 eV. The diffraction measurements were performed in shutterless mode with an angular step size of  $0.1^{\circ}$  and a counting time of 0.1s per frame. The detector geometry parameters were calibrated with *PyFAI* (Kieffer & Wright, 2013) using a powder pattern of the NIST 660b standard LaB<sub>6</sub>. Experimental data were collected using the *Pylatus* software (Dyadkin *et al.*, 2016) and treated using the *SNBL ToolBox* (Dyadkin *et al.*, 2016) and *CrysAlisPro* (Rigaku OD, 2019).

The structures were solved using *SHELXT* (Sheldrick, 2018) and a spherical-atom model was refined using *SHELXL-2017* (Sheldrick, 2015). Multipolar refinement was carried out *vs*. all reflections  $F^2$  with program package *MoPro* (Jelsch *et al.*, 2005). Metal and halogen atoms were modelled as hexadecapoles, O, N and C as octupoles and hydrogens as dipoles; loose

restraints were used for multipoles and exponential  $\kappa$  coefficients of chemically equivalent atoms. Vibrations of metal and halogen atoms were refined as anharmonic using fourth-order Gram-Charlier coefficients. Anisotropic parameters for hydrogen atoms were calculated by the *SHADE3* server (Madsen, 2006) and kept fixed in the multipolar atom refinement; aromatic C–H bond lengths were restrained to 1.077(2) Å and methyl C–H to 1.083(2) Å.

Due to a limited quality of the diffraction data and twinning, **tetrazol** was refined only by spheric model with resolution cut off to 0.80 Å.

For **Me2Py** and **dabco** resolution was cut off at 0.50 Å. In **Me2Py** vibrations of atoms N2B, C8B, N7, C14 and C25 were refined as anharmonic using fourth-order Gram-Charlier coefficients. In **dabco** vibrations of atoms C14A, and C16A were refined as anharmonic using fourth-order Gram-Charlier coefficients.

Geometry and charge-density calculations and analysis of Hirshfeld surfaces were performed by *MoPro* (Jelsch *et al.*, 2005); molecular graphics were prepared using *MoProViewer* (Guillot, 2012) and *CCDC-Mercury* (Macrae *et al.*, 2020). Crystallographic and refinement data are shown in Table 1.

Compound	tetrazol	Me2Py	dabco	quinone*
Empirical formula	$C_{74}H_{42}N_{20}$	$C_{61}H_{31}N_{19}$	$C_{78}H_{59}N_{25}$	$C_{21}H_{11}Cl_5CuIN_3O_2$
Formula wt. / g $mol^{-1}$	1211.28	1029.97	1346.40	705.03
Crystal dimensions / mm	0.08 x 0.02 x 0.01	0.12 x 0.10 x 0.06	0.14 x 0.11 x 0.05	0.12 x 0.06 x 0.04
Space group	$P2_{1}/c$	$P \bar{1}$	$P \overline{1}$	$P 2_1/n$
<i>a</i> / Å	29.0602(6)	13.1638(2)	7.8028(1)	9.90659(16)
<i>b</i> / Å	11.5161(3)	13.5312(2)	13.3989(1)	21.6589(4)
<i>c</i> / Å	19.7924(6)	15.6098(2)	16.5465(1)	17.4384(3)
α / °	90	105.985(1)	80.910(1)	90
eta / °	109.859(3)	101.257(1)	81.4170(1)	140.309(3)
γ/°	90	103.275(1)	83.9710(1)	90
Ζ	4	2	2	4
V / Å <sup>3</sup>	6229.8(3)	2500.30(6)	1683.175(17)	2389.58(7)
$D_{ m calc}$ / g cm <sup>-3</sup>	1.291	1.369	1.329	1.987

**Table 1** Crystallographic, data collection and charge-density refinement details.

$\mu$ / mm <sup>-1</sup>	0.053			2.830
heta range / °	1.52 - 40.67	1.47 – 40.49	1.46 - 40.61	1.48 - 35.51
T / K	100(2)	100(2)	100(2)	100(2)
Radiation wavelength	0.56356	0.56356	0.56356	0.56356
Detector type	Dectris Pilatus3 X 2M	Dectris Pilatus3 X 2M	Dectris Pilatus3 X 2M	Dectris Pilatus3 X 2M
Range of $h, k, l$	-66 < h < 66;	-30 < <i>h</i> <	-18 < h < 17;	-19 < h < 19;
	-26 < k < 26;	30;	-30 < k < 30;	-43 < k < 43;
	-45 < l < 45	-26 < k < 26;	-37 < <i>l</i> < 38	-34 < <i>l</i> < 34
		-35 < <i>l</i> < 35		
Reflections collected	439755	716861	253550	219444
Independent reflections	11124	36397	26904	19889
Reflections with $I \ge 2\sigma$	10370	27031	25613	15514
Absorption correction	multi-scan	multi-scan	multi-scan	multi-scan
$T_{\min}, T_{\max}$	multi-scan	0.891; 1.000	0.7422; 1.0000	0.6147; 1.0000
R <sub>int</sub>	0.0758	0.0424	0.0288	0.0983
Spheric refinement				
Weighting scheme	$w = 1/[\sigma^{2} (F_{o}2) + (0.1701P)^{2} + 8.6629P]$ where P = $(F_{o}^{2} + 2F_{c}^{2})/3$	$w = 1/[\sigma^{2} (F_{o}2)+(0.1 533P)^{2}]$ where $P = (F_{o}^{2}+2F_{c}^{2})/3$	$w = 1/[\sigma^{2}$ (F <sub>o</sub> 2)+(0.0866P) <sup>2</sup> +0.605P] where P = (F <sub>o</sub> <sup>2</sup> +2F <sub>c</sub> <sup>2</sup> )/3	$w = 1/[\sigma^{2} (F_{o}2)+(0.0516P)^{2}+1$ 3.9368P] where P = $(F_{o}^{2}+2F_{c}^{2})/3$
R(F)	0.0696	0.0752	0.0458	0.0864
$R_{\scriptscriptstyle W}\left(F^2 ight)$	0.2250	0.2740	0.1398	0.2176
Goodness of fit	0.851	1.035	1.008	0.983
H atom treatment	Constrained isotropic	Constraine d isotropic	Constrained isotropic	Constrained isotropic
No. of parameters	848	733	930	298
No. of restraints	0	9	9	0
$\Delta ho_{ m max}$ , $\Delta ho_{ m min}$ , $\Delta ho_{ m rms}$ (e Å <sup>-3</sup> )	0.499; -0.384; 0.066	0.970; -1.298; 0.097	1.050; -1.050; 0.070	5.09; -5.32: 0.21

Multipolar refinement			
Weighting scheme	-	$w = 1/[\sigma^2(F_0^2)]$	$w = 1/[\sigma^2(F_0^2)]$
R(F)	-	0.0325	0.0216
$R_w(F^2)$	-	0.0645	0.0545
Goodness of fit	-	1.150	1.491
H atom treatment	-	Constraine d anisotropic	Constrained anisotropic
No. of parameters	-	2141	2860
No. of restraints	-	1112	1351
$\Delta ho_{ m max}$ , $\Delta ho_{ m min}$ , $\Delta ho_{ m rms}$ (e Å <sup>-3</sup> )	-	0.458; -0.609; 0.047	0.296; -0.283; 0.043

\* The best data set so far. Two data sets still have to be integrated.

# Crystal structure of *tetrazol*

Asymmetric unit of **tetrazol** (Fig. 1) containts two symmetry-independent tetrazolium cations and four symmetry-independent partially charged 7,7,8,8-tetracyanoquinodimethane (TCNQ) radical anions. From the stoichiometry, charge of TCNQ moieties is -1/2.



**Figure 1** ORTEP drawing of asymmetric unit of **tetrazol** with atom labelling scheme. Displacement ellipsoids are drawn for the probability of 50 % and hydrogen atoms are shown as spheres of arbitrary radii.

## Crystal structure and charge density of Me2Py

Asymmetric unit of **Me2Py** (Fig. 2) containts a 1,1'-methylenedipyridinium cation, four symmetry-independent partially charged 7,7,8,8-tetracyanoquinodimethane (TCNQ) radical anions and a solvent acetonitrile molecule. From the stoichiometry, charge of TCNQ moieties is -1/2.



**Figure 2** ORTEP drawing of asymmetric unit of **Me2Py** with atom labelling scheme. Displacement ellipsoids are drawn for the probability of 50 % and hydrogen atoms are shown as spheres of arbitrary radii.





**Figure 3** Residual density in the mean plane of four symmetry-independent TCNQ moieties in **Me2py** with all reflections used (left) and only low-angle reflections (sin  $\theta / \lambda < 0.7 \text{ Å}^{-1}$ ) (right) used. Positive density is shown in blue and negative in red; yellow dotted lines represent zero density. The spacing between contours is of 0.05 eÅ<sup>-3</sup>.





**Figure 4** Residual density in the mean plane of two pyridinium rings of the cation in **Me2py** with all reflections used (left) and only low-angle reflections ( $\sin \theta / \lambda < 0.7 \text{ Å}^{-1}$ ) (right) used. Positive density is shown in blue and negative in red; yellow dotted lines represent zero density. The spacing between contours is of 0.05 eÅ<sup>-3</sup>.





**Figure 5** Left: deformation density maps in the mean planes of four symmetry-independent TCNQ moieties in **Me2py**. The spacing between contours is of 0.05 e Å<sup>-3</sup>; the positive density is blue, the negative is red and the zero contour is drawn as a yellow dotted line. Right: Laplacians of electron density in the mean planes of four symmetry-independent TCNQ moieties in **Me2py**. The contours are drawn for 2, 4,  $8 \cdot 10^n$  e Å<sup>-5</sup>, n = -3...2; positive Laplacian is blue and negative is red.





**Figure 6** Left: deformation density maps in the mean planes of two pyridinium rings of the cation in **Me2py**. The spacing between contours is of 0.05 e Å<sup>-3</sup>; the positive density is blue, the negative is red and the zero contour is drawn as a yellow dotted line. Right: Laplacians of electron density in the mean planes of two pyridinium rings of the cation. The contours are drawn for 2, 4,  $8 \cdot 10^n$  e Å<sup>-5</sup>, n = -3...2; positive Laplacian is blue and negative is red.

#### Crystal structure and charge density of dabco

Asymmetric unit of **dabco** (Fig. 7) containts two symmetry-independent *N*,*N*'-dimethyl-dabco cations, five symmetry-independent partially charged 7,7,8,8-tetracyanoquinodimethane (TCNQ) radical anions and a solvent acetonitrile molecule. From the stoichiometry, charge of four TCNQ moieties is -1/2, while the fifth one is neutral. With 103 symmetry-indepentent non-hydrogen atoms (and a total of 162 atoms), **dabco** may be the largest structure refined by unconstrained multipolar model so far.



**Figure 7** ORTEP drawing of asymmetric unit of **dabco** with atom labelling scheme. Displacement ellipsoids are drawn for the probability of 50 % and hydrogen atoms are shown as spheres of arbitrary radii.





**Figure 8** Residual density in the mean plane of five symmetry-independent TCNQ moieties in **dabco** with all reflections used (left) and only low-angle reflections (sin  $\theta / \lambda < 0.7 \text{ Å}^{-1}$ ) (right) used. Positive density is shown in blue and negative in red; yellow dotted lines represent zero density. The spacing between contours is of 0.05 eÅ<sup>-3</sup>.





**Figure 9** Left: deformation density maps in the mean planes of five symmetry-independent TCNQ moieties in **dabco**. The spacing between contours is of 0.05 e Å<sup>-3</sup>; the positive density is blue, the negative is red and the zero contour is drawn as a yellow dotted line. Right: Laplacians of electron density in the mean planes of five symmetry-independent TCNQ moieties in **dabco**. The contours are drawn for 2, 4,  $8 \cdot 10^n$  e Å<sup>-5</sup>, n = -3...2; positive Laplacian is blue and negative is red.

#### Crystal structure of quinone

Asymmetric unit of **quinone** (Fig. 10) comprises a molecule of a complex [CuICl(terpy)] (terpy = 2,2';6',2"-terpyridine) and a molecule of tetrachloroquinone.

So far, only four of six data sets have been processed and integrated. Spherical refinement of the best of them (Fig. 10) yielded *R* value of 8.64 % and maximum residual density of 5.09 e Å<sup>-3</sup> using all data. With data cut off to atomic resolution, the maximum residual density is reduced below 2.0 e Å<sup>-3</sup>, so the atomic resolution structure is publicable.

Multipolar refinement has not been completed, but so far it yielded unacceptably high residual density around the I1 atom of 5.07 e Å<sup>-3</sup>. It will probably be improved, however, it is still not certain that the charge density would eventually be publicable.



**Figure 10** ORTEP drawing of asymmetric unit of **quinone** with atom labelling scheme. Displacement ellipsoids are drawn for the probability of 50 % and hydrogen atoms are shown as spheres of arbitrary radii.

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