



	Experiment title: EXAFS study of the local structure of luminescent erbium centers in titanosilicates	Experiment number: CH884
Beamline: BM29	Date of experiment: from: 2/2/00 to: 4/2/00	Date of report: 12/7/00
Shifts: 6	Local contact(s): Michael Borowski	<i>Received at ESRF:</i> 28 JUL. 2000
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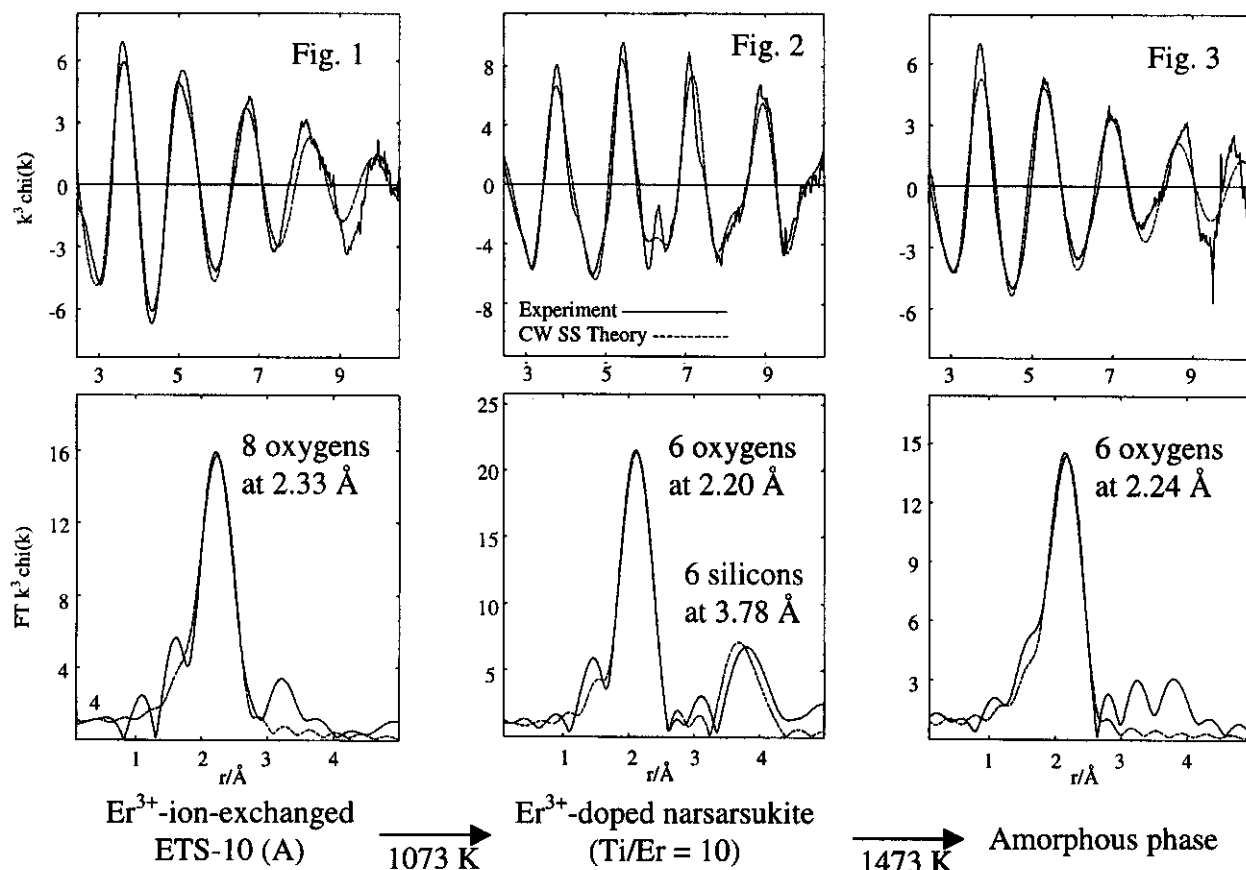
Aims. The microporous titanosilicate ETS-10 $[(\text{Na},\text{K})_2\text{TiSi}_5\text{O}_{13}\cdot x\text{H}_2\text{O}]$ doped with different concentrations of Er^{3+} ions may be used as a precursor for preparing novel dense materials, analogues of the mineral narsarsukite $[(\text{Na},\text{K})_2\text{TiSi}_4\text{O}_{11}]$, which display a high and stable room-temperature emission in the visible and infrared spectral regions [J. Rocha *et al.*, *J. Mater. Chem.*, 2000, **10**, 1371]. Er L_{III} -edge EXAFS studies were carried out in order to probe the local erbium environment in these materials and gain a better understanding of their unusual optical properties.

Experimental conditions. Er L_{III} -edge X-ray absorption spectra were collected at room temperature (ca. 10^{-5} mbar) and/or 50 K (Oxford cryostat, He exchange gas) in transmission mode on BM29 at the ESRF, operating at 6 GeV in 2/3 filling mode with typical currents of 170–200 mA. [double crystal Si(311) monochromator, ion chamber detectors]. Spectra were recorded in the range 8.058–9.3 KeV (50% harmonic rejection, EXAFS scan with optimised energy mesh, 3 s per scan point). Powdered samples were diluted with appropriate amounts of boron nitride and 13mm diameter pellets prepared. A total of about 40 complete EXAFS scans were acquired for 11 different samples (r.t. and/or 50 K for each sample, 1-2 scans per sample). A reference sample of bulk Er_2O_3 was used as a model standard for determining coordination numbers and interatomic distances.

Analysis. Background subtraction and EXAFS curve-fitting analyses were carried out within the programs EXBACK and EXCURVE (version EXCURV98, Daresbury Laboratory, U.K.).

Sample characteristics. EXAFS spectra were measured for Er^{3+} -ion-exchanged ETS-10 samples calcined at 973 K, 1073 K, 1273 and 1473 K. Er^{3+} -doped narsarsukite crystallises at 973–1173 K, at ca. 1190 K the materials melt and at ca. 1473 K an amorphous phase is obtained. Two ETS-10 samples loaded with high (A) and low (B) amounts of Er were used that led to Er^{3+} -doped narsarsukite with Ti/Er = 10 and 23 respectively (ca. 4 and 2 mass% erbium).

Results. EXAFS spectra were of satisfactory quality up to 11 \AA^{-1} for samples A (edge jump ca. 0.15 a.u.) and up to 9 \AA^{-1} for samples B (edge jump ca. 0.05 a.u.). The spectrum of Er^{3+} -ion-exchanged ETS-10 (sample A, 50 K, Fig. 1) was adequately fitted by a single shell of 8 oxygens at 2.33 \AA , presumably comprising at least some hydration water molecules ($2\sigma^2 = 0.026 \text{ \AA}^2$). Er^{3+} -doped narsarsukite, formed at 1073 K, provided a markedly different EXAFS spectrum and Fourier transform (Fig. 2). A well-defined shell of 6 oxygens could be fitted at 2.20 \AA ($2\sigma^2 = 0.008 \text{ \AA}^2$) and an additional shell of 6 silicon atoms at 3.78 \AA ($2\sigma^2 = 0.012 \text{ \AA}^2$). The $\text{Er}\cdots\text{Si}$ distance seems reasonable for $\text{Er}-\text{O}-\text{Si}$ bonding. There was no evidence for $\text{Er}\cdots\text{Er}$ correlations within the short range order, a result especially pertinent to the optical properties of these materials. The average $\text{Er}-\text{O}$ bond distance is significantly shorter than that for Er_2O_3 (6 O's at 2.26 \AA , $2\sigma^2 = 0.012 \text{ \AA}^2$). These results suggest the existence of $[\text{ErO}_6]$ complexes which are strongly bonded internally and weakly coupled to the narsarsukite lattice. The amorphous sample, formed at 1473 K, provided a different EXAFS spectrum and was found to have 6 nearest oxygen neighbours at 2.24 \AA ($2\sigma^2 = 0.017 \text{ \AA}^2$) but no well-defined second nearest neighbour shell (Fig. 3). Comparable results were obtained with the samples containing a lower amount of erbium, indicating that the local environment around the Er^{3+} centres was the same for differently loaded samples calcined to the same temperature.



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