ESRF	Experiment title: Uranyl speciation in cement phases using XAS	Experiment number: ME-308
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While the speciation of U(VI) in alkaline solution is relatively well known, the interaction of U(VI) with cement and cement phases remains poorly understood. Some authors have reported inner-sphere complexation between uranyl and hydrothermally altered concrete in the pH range 9-12. However the chemical phase responsible for this process has not been identified. Moreover the identification of the U solubility limiting phase in cementitious systems is subject to debates. To deepen our knowledge on the U(VI)/cement system, a XAS study was initiated and measurements at the U L_3 edge (17.17 keV) carried out. Selected XANES (X-ray Absorption Near Edge Structure) and EXAFS (Extended X-ray Absorption Fine Spectroscopy) spectra are presented in Figure 1. For oversaturated conditions with respect to the precipitation of the U-solubility limiting phase(s), the spectra of HCP-Ov are similar to those of the UO₂-Ca reference (synthesized at pH=13.3 in the presence of Ca(II)). Some complementary experiments are on the way to identify UO₂-Ca solid but it most likely corresponds to becquerelite $(Ca[(UO_2)_6O_4(OH)_6].8H_2O)$. For undersaturated conditions, the spectra of HCP-Und display some marked differences with those of UO₂-Ca but are similar to those of CSH-0.7, which may reveal that CSH phase is the U(VI) uptake-controlling phase in cementitious systems. Data fitting are currently undertaken to identify the U(VI) coordination mode in the CSH and HCP solids.

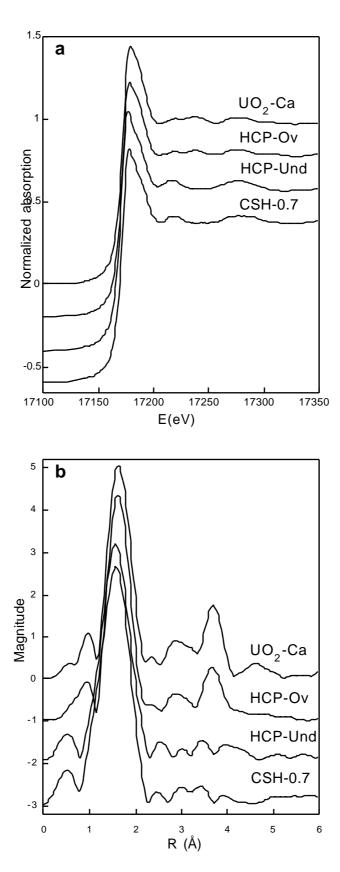


Figure 1: XANES (*a*) and PRDF (*b*) spectra of UO_2 -Ca (reference compound), U(VI) immobilized on HCP (HCP-Ov: Oversaturated conditions; HCP-Und: Undersaturated conditions) and on CSH (CSH-0.7)