



	Experiment title: NaMgF ₃ Perovskite at High Pressures and Temperatures	Experiment number:
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Report:

Introduction: The perovskite with general formula ABX₃, consists of corner-sharing BX₆ octahedra that surround A cations in 12-fold coordination. Interest in this unique structure stems, in part, from the central role of MgSiO₃ in Earth Science. This composition, in the perovskite structure, is believed to constitute the majority component in the lower mantle, and unexpected lateral variations in seismic velocities in this region (Woodhouse and Dziewonski, 1989) might result from changes in temperature. This explanation is rationalized if dG/dT (G = shear modulus) is 2-4 times larger than dK_T/dT (Yeganeh-Haeri and Weidner, 1989).

Measurements of the structure, properties and phase transitions in MgSiO₃ at the pressure and temperature conditions of the lower mantle are difficult and so the study of a suitable analogue compound at more moderate conditions of P and T can provide insight. NaMgF₃ is orthorhombic and isostructural and isoelectronic with MgSiO₃; It is proposed as an ideal analogue material (O'Keeffe and Bovin, 1979). Previously, NaMgF₃ was shown to undergo a phase transition from orthorhombic (Pbnm) directly to cubic (Pm-3m) at elevated pressure and temperature with energy-dispersive x-ray diffraction, while dK/dT and $d\alpha/dP$ were both found to be 0 (Zhao et al., 1994). A value of zero for both dK/dT and $d\alpha/dP$ implies that compression and heating each have separate and different effects on sample structure. Zhao et al. (1994) suggest that changes in temperature tilts MgF₆ octahedra, while changes in pressure compress Mg-F bonds. We are interested in using angle dispersive x-ray diffraction to examine the effect of simultaneous pressure and temperature on: thermal expansion and compressibility, rotation of MgF₆ octahedrons, compression of Mg-F bonds, and the slope of the orthorhombic / cubic phase transition.

Methods and Materials: The polycrystalline sample of NaMgF₃ was prepared in air using conventional solid-state methods. Samples were mixed with 25 mol% B₄C to insure proper grain averaging throughout the experiment. Data collection was performed in angle-dispersive mode using a MAR imaging plate. Simultaneous EOS of BN and Au was used to determine pressure and temperature, further details of the experimental setup can be found elsewhere (Crichton and Mezouar, 2002).

Results: A total of 331 angle dispersive x-ray diffraction patterns were collected during beamtime. Our analysis confirms that NaMgF₃ transforms from orthorhombic directly to cubic at all pressures and temperatures examined (up to 6 GPa and 1400°C). Also, we find a much steeper slope for the phase transition than was found previously 130 vs. 45 K/GPa (Zhao et al., 1994). This discrepancy may be a result of poor grain averaging in the Zhao (1994) study, as we find that without an inert secondary phase to buffer contact between grains, NaMgF₃ readily and quickly consolidates crystallites at high temperature. A preliminary second-order fit of the Birch-Murnaghan equation of state to data yields the following constants: $K_0 = 59.4 \pm 1.1$ GPa, $\alpha_0 = 9.23 \pm .58$ K⁻¹, and $dK/dT = -0.016 \pm .0039$ GPa•K⁻¹. The effect of pressure and temperature on atomic positions and bond lengths using Rietveld refinement is currently ongoing.

Conclusions: With the preliminary finding that compressibility of NaMgF₃ changes substantially with temperature, we are able to conclude (using NaMgF₃ as an analogue and assuming lateral temperature variations co-occur with large lateral $d\ln V_s/d\ln V_p$ values) that values of dG/dT for MgSiO₃ must be anomalously large relative to other lower mantle silicate and oxide phases to be ultimately responsible for lateral variations in seismic velocities. One must then examine the effect of perovskite phase transitions and/or regions of melt as a possible explanation.

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