HELMHOLTZ ZENTRUM DRESDEN ROSSENDORF ROBL-CRG	Experiment title: U and Np sorption on bacteria and biogenically formed ferihydrite	Experiment number: 20-01-763
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Report:

XANES (Fig. 1a) and EXAFS spectra (Fig. 1b and c) of the U sorped biogenic ferrihydrite sample are shown along with selected references. The XANES position and especially its fine structure are indicative of the hexavalent uranyl moiety. The "yl-shoulder" at about 17.175 keV is fully expressed, hence gives no indication for a significant amount of U^{IV}. The k³-weighted chi-spectrum and its Fourier transform magnitude bears close resemblance to the bidentate edge-sharing innersphere sorption complex (¹E), which is the main sorption species on ferrihydrite. ¹ Due to the CO₂-feeding of the *Gallionella* culture, the biogenic ferrihydrite precipitated in this system may have a relatively high carbonate loading of its surface, potentially causing the formation of a smaller portion of the type-B ternary uranyl –carbonato sorption complex determined for high pH

and high p_{CO2} . ¹ To elucidate this potential second sorption species, we performed iterative target test factor analysis (ITFA) using the spectra of the two endmember species ². The good reproduction of the *Gallionella* spectrum along with those of the two endmember spectra by 2 principal components (compare the black and blue lines in Fig. 1b and c) proves that the *Gallionella* samples contain no third sorption species. Using the iterative target test modul of the ITFA software package, we determined that the ¹E complex is in fact predominant with 95%, while the ternary uranyl-carbonato complex is present only to 5%.

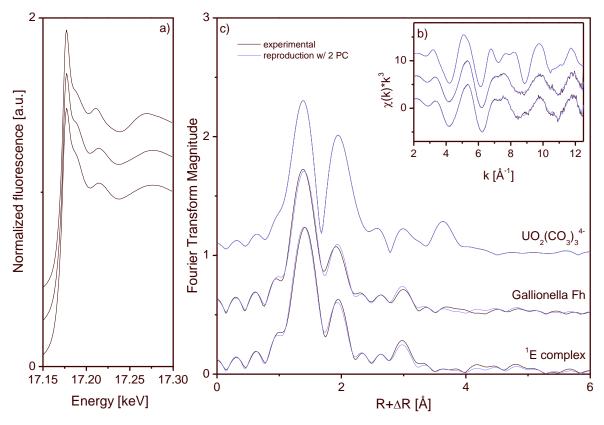


Figure 1. U-L_{III} XAS spectra of uranium sorption sample on Gallionella-produced ferrihydrite, together with the limiting species observed for ferrihydrite sorption samples, i.e. a bidentate edge-sharing uranyl sorption complex (${}^{1}E$) and the aqueous triscarbonato complex ($UO_{2}(CO_{3})_{3}^{4-}$) representative for a type-B ternary uranyl carbonato sorption complex taken from Rossberg et al. (2009) ¹

- (1) Rossberg, A. et al. *Environ. Sci. Technol.* **2009**, *4*3 (5), 1400–1406.
- (2) Rossberg, A.et al. Anal. Bioanal. Chem. 2003, 376 (5), 631-638.