



## Experiment Report Form



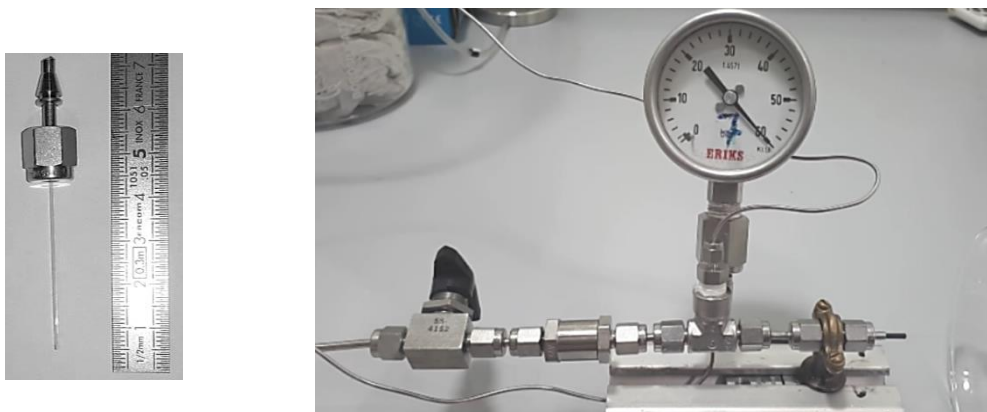
	<b>Experiment title:</b> Enhanced formation of CH <sub>4</sub> clathrate hydrates in hydrophobic confinement	<b>Experiment number:</b> 26-02-868
<b>Beamline:</b> BM26B	<b>Date of experiment:</b> from: 05/04/2018 to: 09/04/2018	<b>Date of report:</b> 26/04/2018
<b>Shifts:</b> 12	<b>Local contact(s):</b> Daniel Hermida Merino	<i>Received at ESRF:</i>
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### Report:

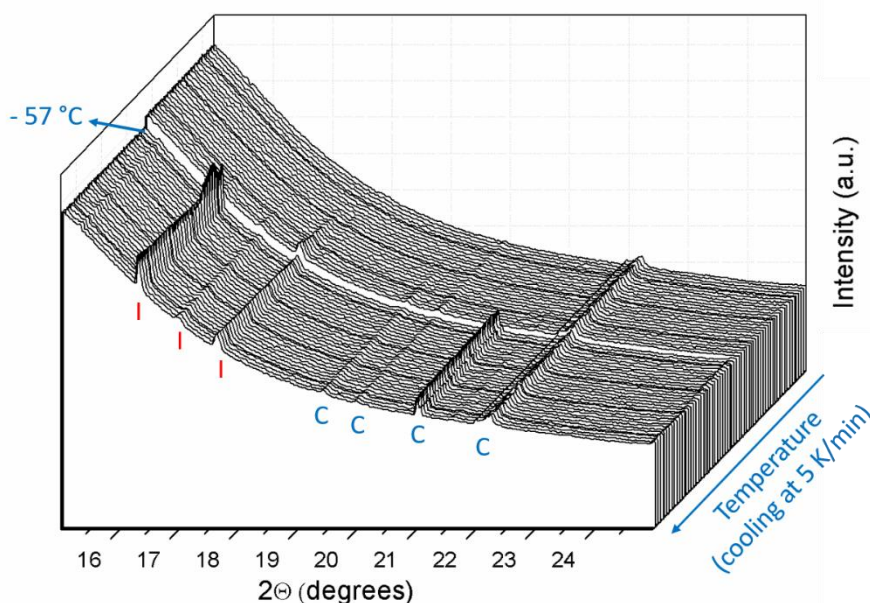
Over the past decade, considerable scientific effort has been devoted to the formation of CH<sub>4</sub> clathrate hydrates. Clathrate hydrates can crystallize when non-polar gases and water are equilibrated under high pressure at low temperature. . Van der Waals interactions between gas molecules and the water force the latter to reorganize its hydrogen bonding network into ice-like materials with cavities large enough to store a variety of gas molecules such as CH<sub>4</sub> and CO<sub>2</sub>. As 1 m<sup>3</sup> of CH<sub>4</sub> hydrate can store up to 160 m<sup>3</sup> of gas, clathrate hydrates offer a safe and sustainable way of storing large amounts of non-polar gaseous energy carriers such as CH<sub>4</sub> and are expected to play a key role in the energy economy of the future.<sup>1</sup>

During this stay at the DUBBLE beamline, CH<sub>4</sub> clathrate hydrate crystallization was monitored with *in situ* SAXS/WAXS to investigate the kinetics of clathrate hydrate formation. X-ray diffraction has been applied to study CH<sub>4</sub> clathrate hydrate formation in the micropores of carbonaceous host materials, but the effect of different pore sizes (1-20 nm) and surface chemistry (siliceous materials with varying hydrophobicity) remains poorly understood.<sup>2-4</sup> Reverse phase microporous silica gels ( pore Ø6nm) hydrophobized with C<sub>8</sub> and C<sub>18</sub> alkyl groups and POSISil materials, a specific type of porous silicone materials with general formula (R<sub>2</sub>SiO)<sub>n</sub> combining micro-, meso- and macropores, were used as host material. Experiments were carried out in 1 mm quartz capillaries containing defined amounts of the nanoporous host materials and water. Using our

unique high-pressure set-up, which allows leak-proof connection between a capillary and virtually any instrument having Swagelok fittings (Figure 1), the capillaries were pressurized with CH<sub>4</sub> up to pressures of 60 bar and the SAXS/WAXS profile was measured as function of temperature. The results with C<sub>8</sub> silica gels look promising, already evidencing the formation of CH<sub>4</sub> clathrate hydrate from hypercooled water (Figure 2). The remaining data is still being analyzed.



**Figure 1.** Left: 1 mm quartz capillary which has been modified to allow connection to a pressure regulating system with standard Swagelok components. Right: Pressurization of the capillary with 60 bar of CH<sub>4</sub>.



**Figure 2.** WAXS diffractogram showcasing the formation of CH<sub>4</sub> clathrate hydrate inside the pores of a hydrated microporous silica grafted with C<sub>8</sub> alkyl groups. Reflections characteristic of CH<sub>4</sub> hydrate structures are denoted by 'C', whereas reflections attributed to the presence of hexagonal ice are indicated with 'I'. The onset of clathrate formation is clearly visible at -57 °C.

- (1) Casco, M. E.; Silvestre-Albero, J.; Ramirez-Cuesta, A. J.; Rey, F.; Jorda, J. L.; Bansode, A.; Urakawa, A.; Peral, I.; Martinez-Escandell, M.; Kaneko, K.; Rodriguez-Reinoso, F. *Nat Commun* **2015**, *6*, 6432.
- (2) Iiyama, T.; Nishikawa, K.; Suzuki, T.; Kaneko, K. *Chemical Physics Letters* **1997**, *274*, 152-158.
- (3) Futamura, R.; Ozeki, S.; Iiyama, T. *Carbon* **2015**, *85*, 8-15.
- (4) Liu, X.; Zhou, L.; Li, J.; Sun, Y.; Su, W.; Zhou, Y. *Carbon* **2006**, *44*, 1386-1392.